

The Macrominerals That Give us Grief Ca, P, Mg

How they are absorbed, the problems that arise
when they are not, and strategies to avoid these
problems

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Ames, IA

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Jesse Goff - Conflicts of Interest

Invented **Soychlor** anionic supplement – former advisor for the Landus Farmer's Cooperative on issues related to formulating diets with Soychlor.

Designed and Tested *Solanum glaucophyllum* boluses to help maintain normal blood Calcium in fresh cows, now marketed by Silberhorn Animal Health as Goff-Bol.

Owner of GlycoMyr- a company that makes a vitamin drench for baby pigs. Working to utilize plant sources of active vitamin D compounds for human inflammatory bowel disease

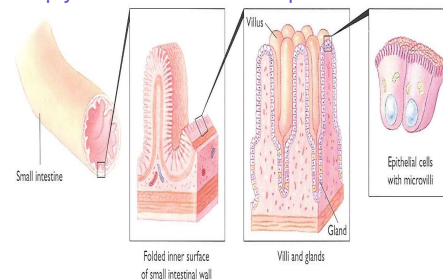
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Anatomy of the Absorptive Epithelium of the GI Tract

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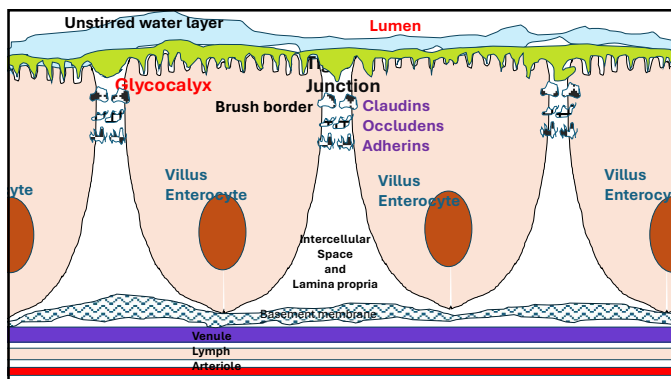
Intestine

Folds, villi, crypts and especially the microvilli (brush border)
greatly multiply surface area for absorption



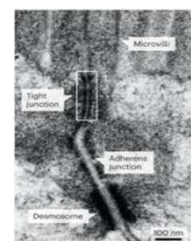
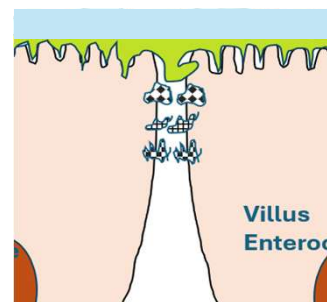
Sjaastad, Hove, Sand: Physiology of Domestic Animals

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Tight Junctions are not completely tight



Horowitz et al., Nature Reviews 2023

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Tight Junctions and Intestinal Permeability

Resistance of intestine epithelia TIGHT JUNCTIONS to "leakage" ~ 6 ohms. (bladder wall resistance is 70,000 ohms!)

Absorption of minerals across the tight junctions between cells is called **PARACELLULAR ABSORPTION**.

Occurs ONLY if lumen IONIZED concentration of a mineral is many fold higher than on the interstitial side of the tight junction. **THEREFOR NOT THE MAJOR ABSORPTION PATHWAY FOR MOST MINERALS**

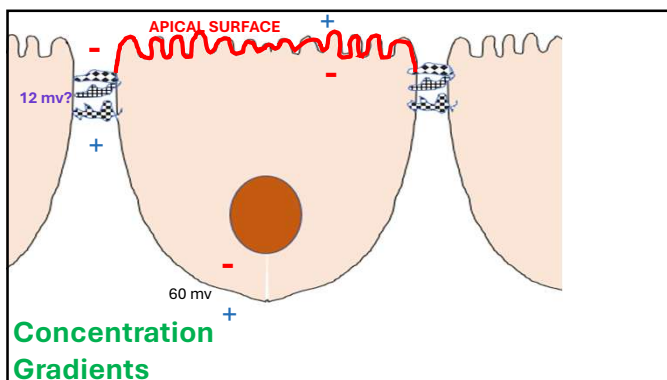
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Transcellular absorption of Minerals

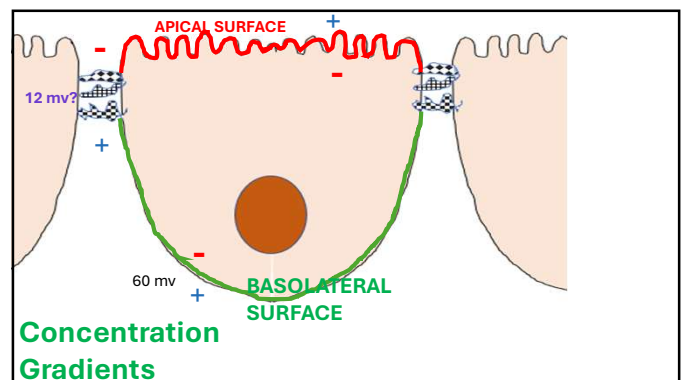
3 Components

1. Specific transporter / channel to allow charged, hydrophilic Mineral to cross lipid bilayer of **apical membrane** of enterocyte.
2. Carrier proteins to carry Mineral across interior of cell
- In the free ionized state minerals often act as free radicals or trigger reactions within the cell cytosol
3. Another specific transporter / channel to move the mineral across the **basolateral membrane** and into the interstitial fluid and blood.

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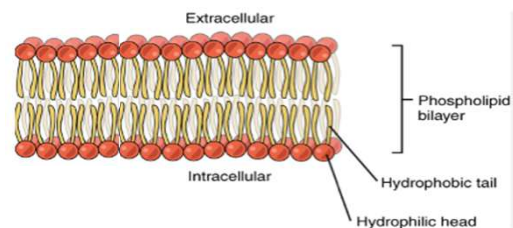
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Phospholipid bilayer blocks passage of charged, hydrophilic minerals across it!

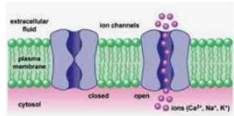


Macro and Trace Minerals are charged particles and NOT Lipid soluble

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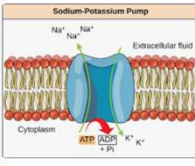
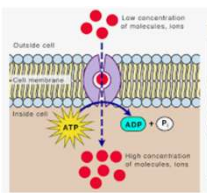
Ion Channel

Allows passive diffusion down an electrical or concentration gradient

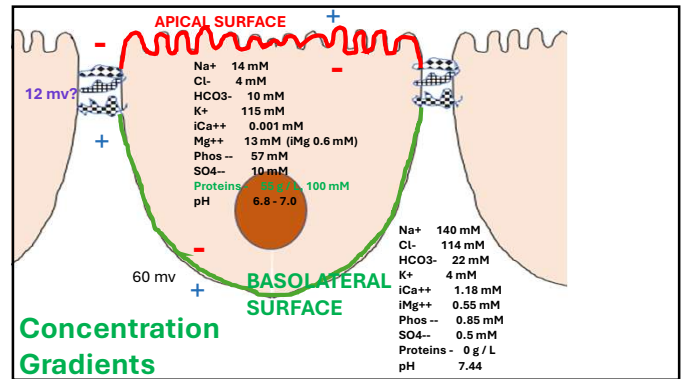


Ion Pumps and Ion exchangers

Energy is required to move an ion against an electrical or concentration gradient
Generally requires energy of ATP or co-transport with an ion that is moving down its electrical and concentration gradient



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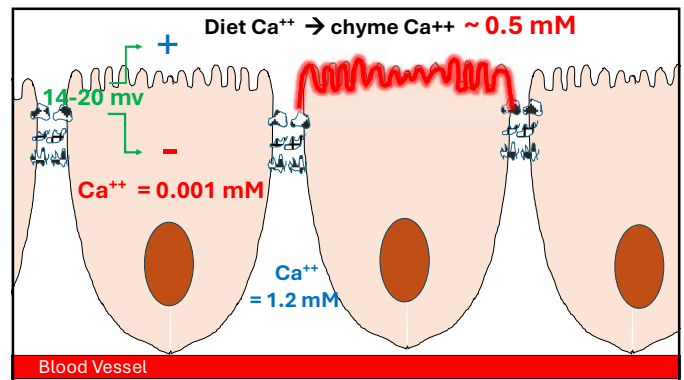
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Electrical and Chemical Gradients must be overcome to move a mineral across an intestinal cell and into the blood.

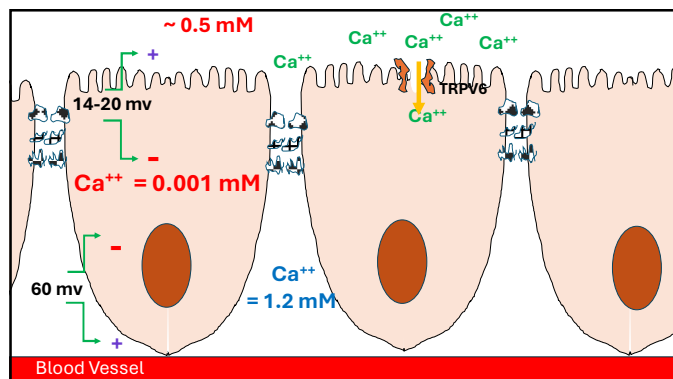
Example - CALCIUM

Ca has a +2 charge when in solution in water or is "ionized". Only ionized Ca is eligible for absorption

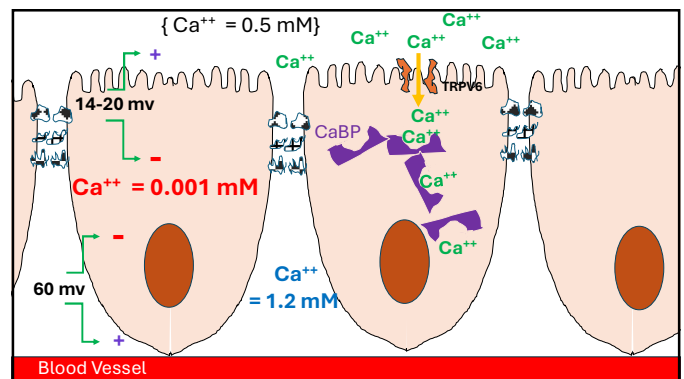
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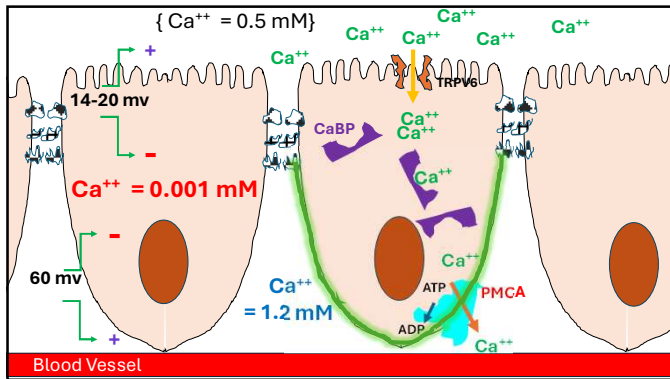
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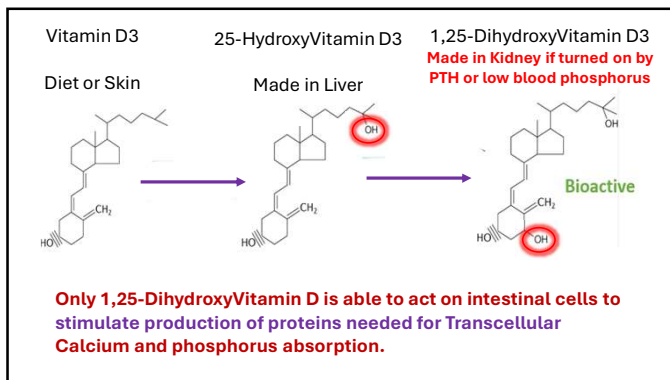
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In Ruminants and Many Monogastric Animals

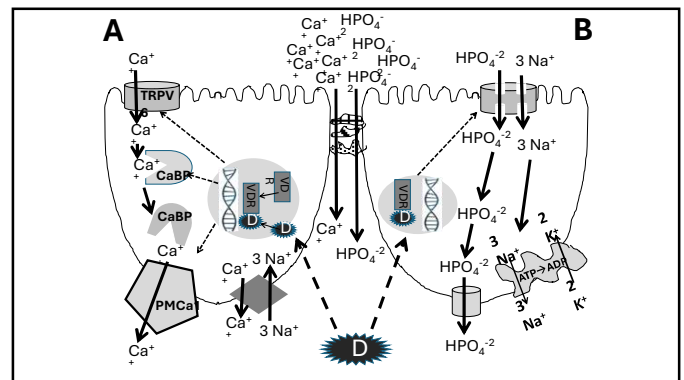
Calcium and Phosphorus (as phosphate HPO_4^{-2}) absorption across cells is **tightly controlled**

The bioactive form of vitamin D, **1,25-dihydroxyvitamin D**, controls the production of Ca and Phosphorus absorbing proteins by the intestinal epithelial cells

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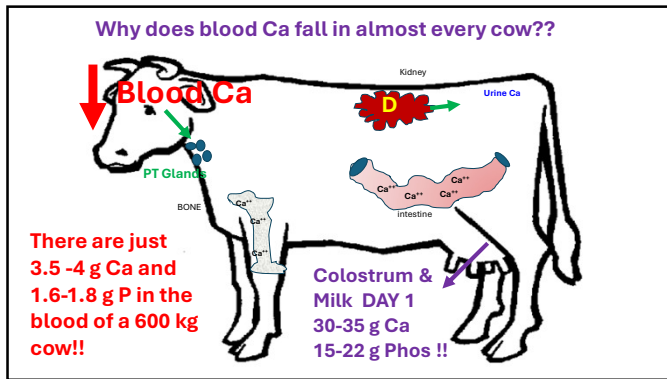
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Milk and Colostrum Macrominerals

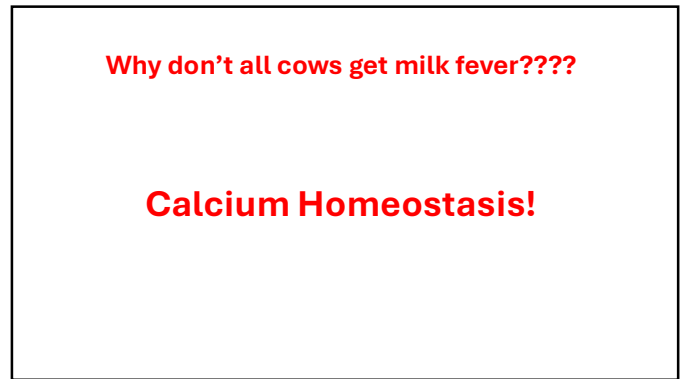
Mineral	Milk at 40 liters/day g mineral /Liter (g / day typical)	Colostrum 7 liters /day Transition milk 10 liters/day		Total g/day 1
		Colostrum g/Liter	Transition milk g/Liter	
Calcium	1.1 (44 g/day)	2.4	1.7	33.8 g
Phosphorus	0.9 (36 g/day)	1.3 *	1	19.1 g
Magnesium	0.12 (4.8 g/day)	0.35	0.2	4.5 g
Sodium	0.6 (24 g/day)	0.9	0.7	13.3 g
Potassium	1.5 (60 g/day)	1.45	1.5	25.2 g
Chloride	1.1 (44 g / day)	1.2	1.2	20.4 g

Adapted from Valdecabres and Silva-del-Rio, 2022; Westhoff et al., 2024

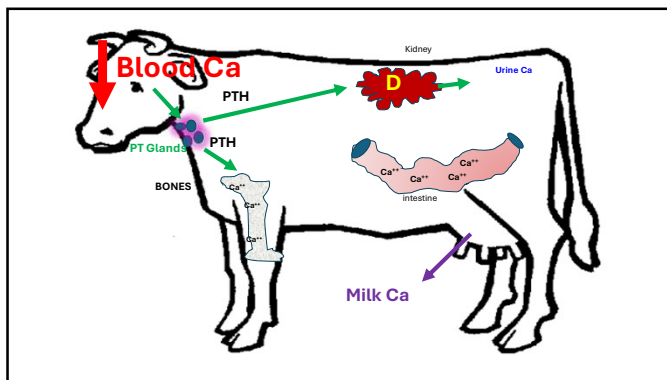
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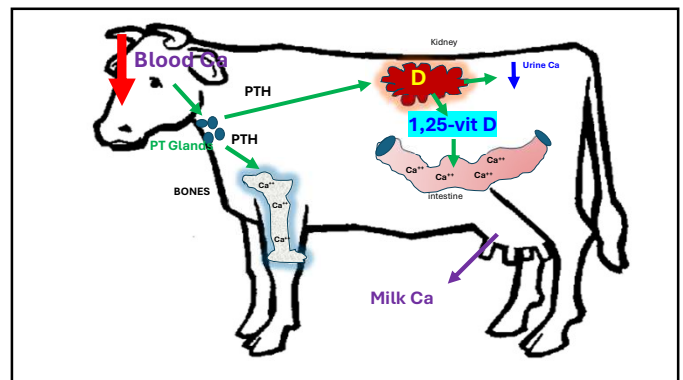
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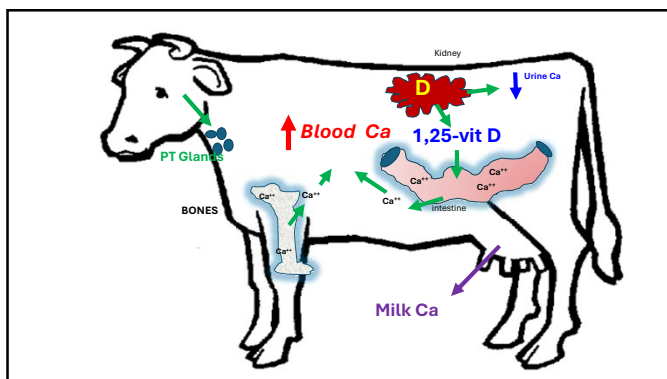
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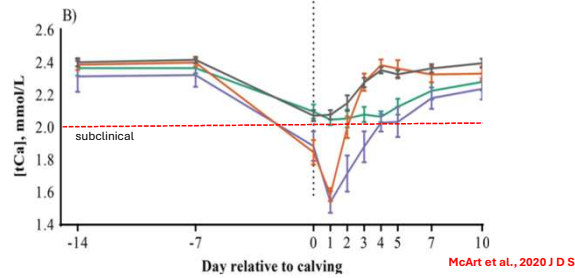


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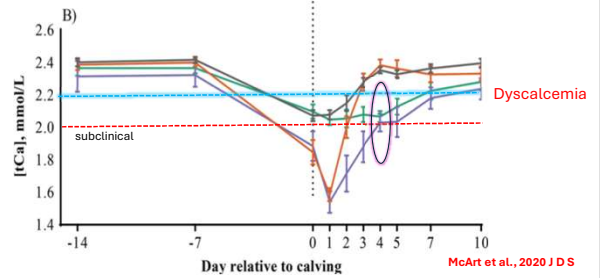
Some cows do not develop any subclinical hypocalcemia (black).
Transient hypocalcemia (red) associated with higher milk production.
Persistent hypocalcemia (purple) associated with higher cull rate.
 Delayed - A few cows develop hypocalcemia after day 2 of lactation (green).



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Dyscalcemia = cows with blood Ca below 8.8 mg/dl (2.2 mM) on day 4.

- reduced DM intake, rumination, fertility
- 1.7 to 5.3 times more likely to be culled or dead
- reduced milk production



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Parity	Ca nadir	% cows with Ca nadir < 8.0	% SCH at 2 DIM	% Dyscalcemia Ca < 8.8 4 DIM	
2	7.96 ± 0.12 ^a	41 ± 6 ^c (122/298)	20 ± 3 ^b (54/272)	27 ± 5 ^b (61/227)	Goff et al, Frontiers Veterinary Science 2025
3	7.48 ± 0.16 ^b	63 ± 6 ^b (123/195)	27 ± 4 ^b (48/176)	41 ± 6 ^a (68/165)	Old Cows have more subclinical hypocalcemia (Ca below 8 mg/dl) and More Dyscalcemia (Ca below 8.8 on day 4)
≥ 4	6.80 ± 0.16 ^c	87 ± 4 ^a (145/167)	45 ± 0.4 ^a (60/134)	41 ± 6 ^a (58/142)	

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Why doesn't Ca Homeostasis work in all cows???

OLD AGE: Cow stops growing by end of 2nd lactation!
Bones are slow to release Ca, heifers don't get milk fever!

Renal production of 1,25-dihydroxyvitamin D slows.

- lower # of receptors for 1,25-dihydroxyvitamin D in intestine of older cows

Failure to produce 1,25-dihydroxyvitamin D quickly enough and in an adequate amount is a major factor contributing to hypocalcemia and Milk Fever

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ALL CLOSE-UP DIETS SHOULD FOLLOW THESE 3 PRINCIPLES

Keep Diet Potassium as low as possible

- Avoid alfalfa and rye/ brome type grass hays or haylages (2.2-4% K)
- Use Prairie hays and clean wheat straw (0.8 – 1.4% K)
- Use some corn silage (0.9-1.3% K)

Keep Diet Phosphorus as low as possible

- Avoid Canola meal (1.05% P), DDG (0.8% P) and Brewers grains (0.65% P)

Provide Adequate Magnesium – Mg needed for PTH secretion and action on target tissues

- > 0.4% Mg for prepartum and early postpartum diets
- Use Readily soluble Mg source

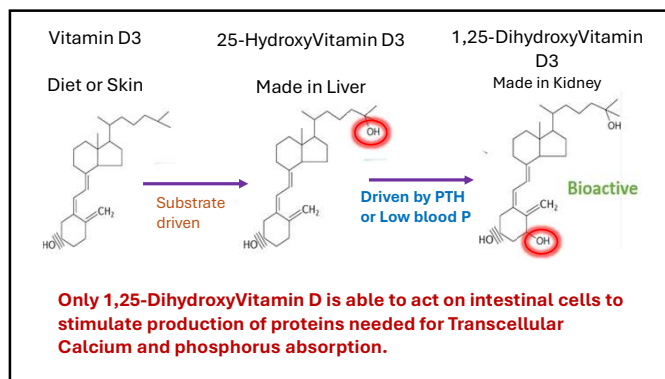
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1. Close-up Diets High in potassium cause alkalosis of blood and urine

Changes shape of receptor for parathyroid hormone so it does NOT RECOGNIZE PTH as well .

Bone and kidney surrounded by lots of PTH, but bone fails to release Ca and kidney fails to produce 1,25-dihydroxyvitamin D.

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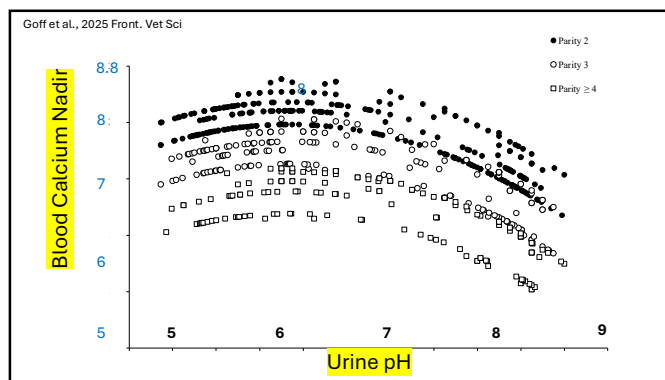


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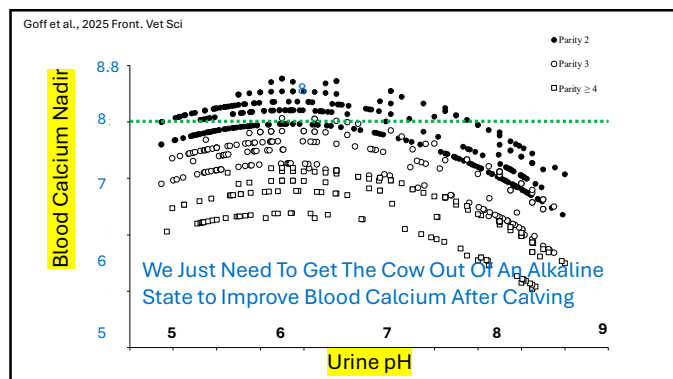
1. Diets High in potassium cause alkalosis → bone & kidney PTH resistance

Anionic diets ↓ alkalinity of blood allowing parathyroid hormone to work on the kidney and increase synthesis of 1,25-dihydroxyvitamin D!

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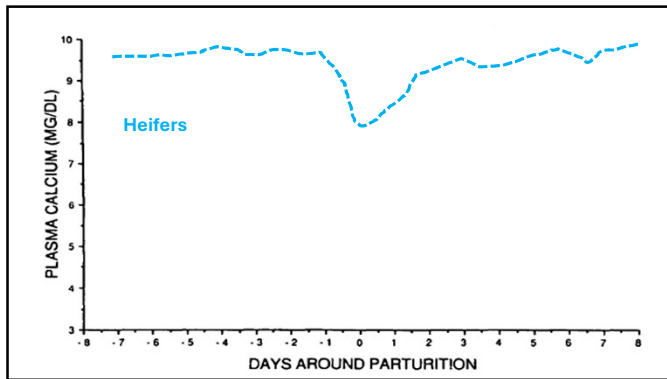
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Blood Ca and blood 1,25-vit D concentrations as heifers and cows go thru the onset of lactation.

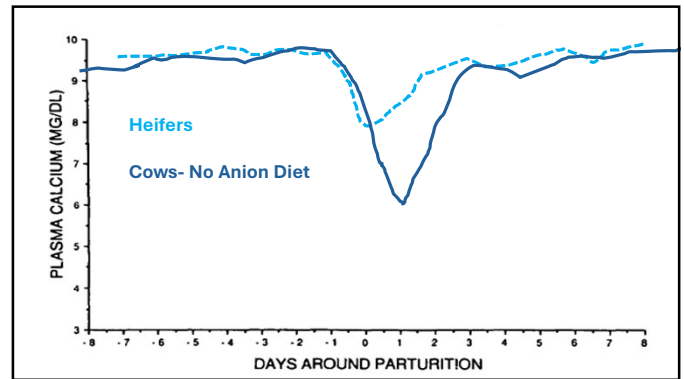
Data from Horst et al, 1977; Goff et al., 1989; Gaynor et al., 1989; Goff and Horst, 1997;

Older cows struggle to produce adequate amounts of 1,25-dihydroxyvitamin D quickly enough to prevent hypocalcemia

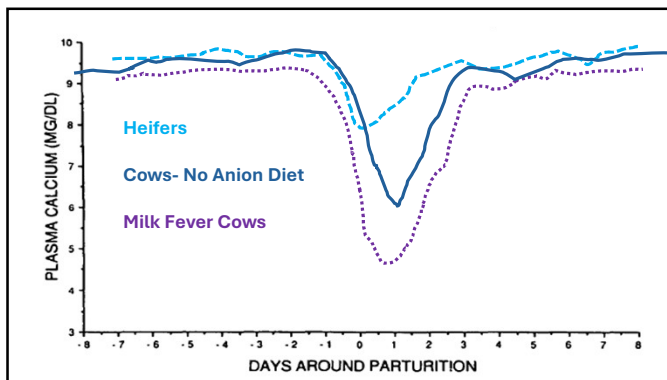
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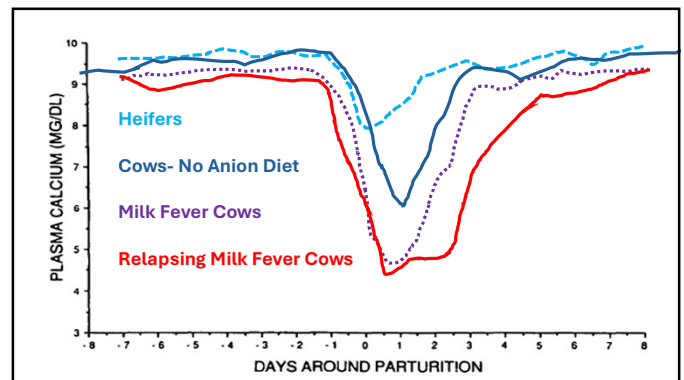
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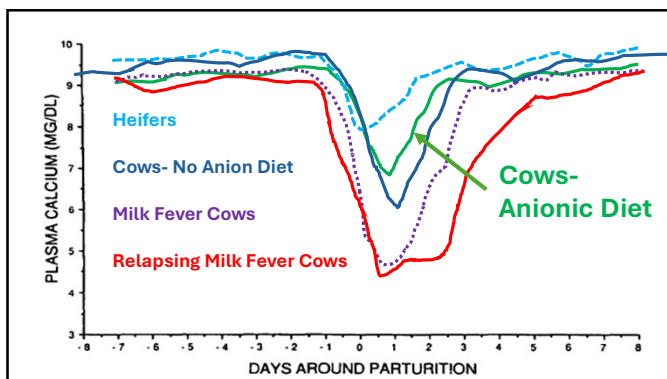
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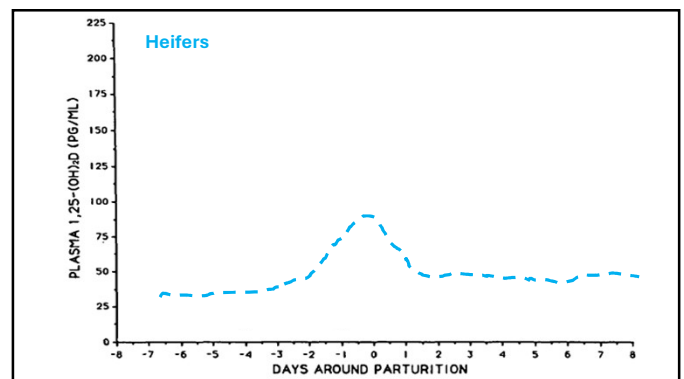
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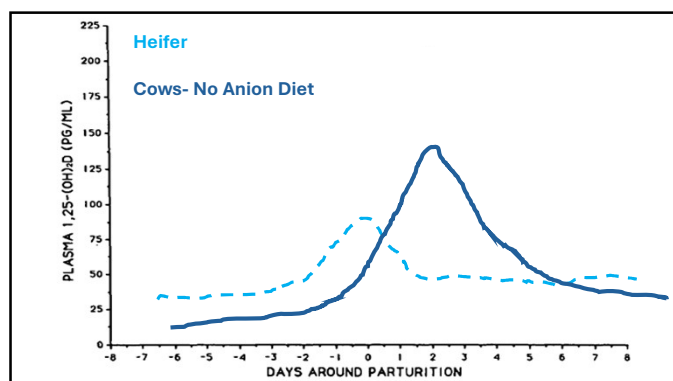
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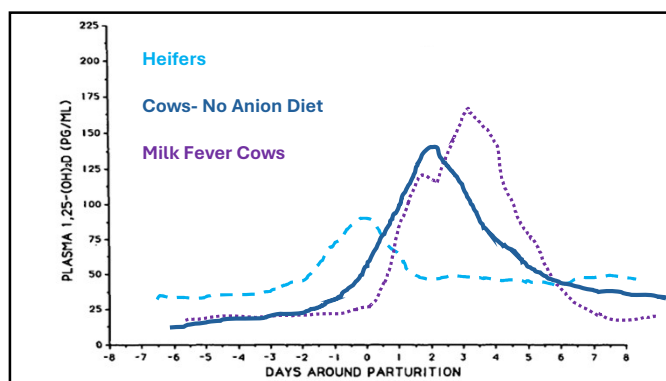
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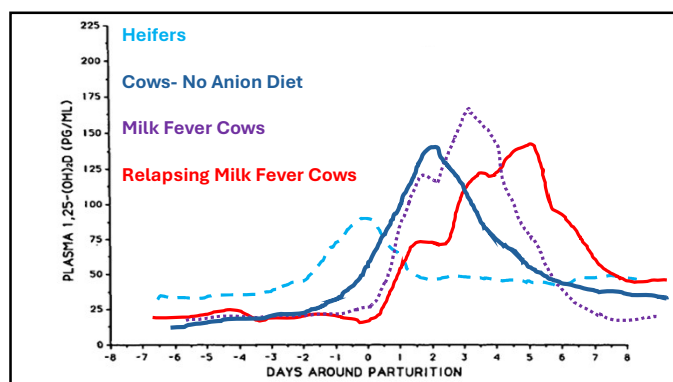
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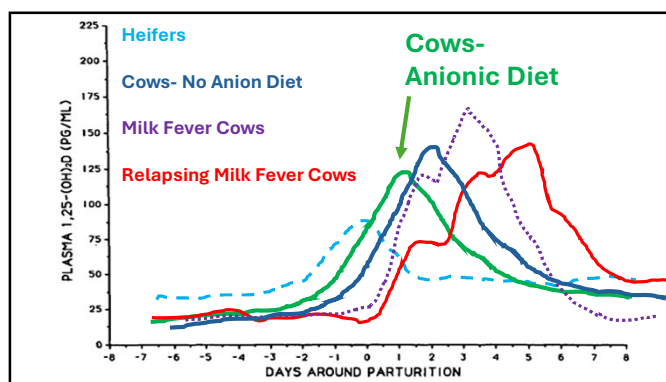
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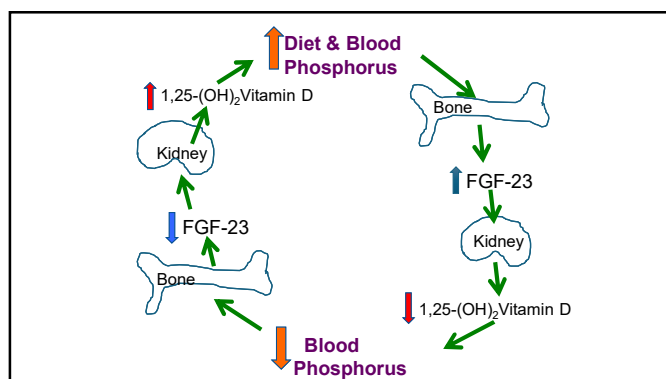


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1. Diets High in **potassium** causing alkalosis → bone & kidney PTH resistance
Anionic diets ↓ alkalinity of blood allowing parathyroid hormone to work!

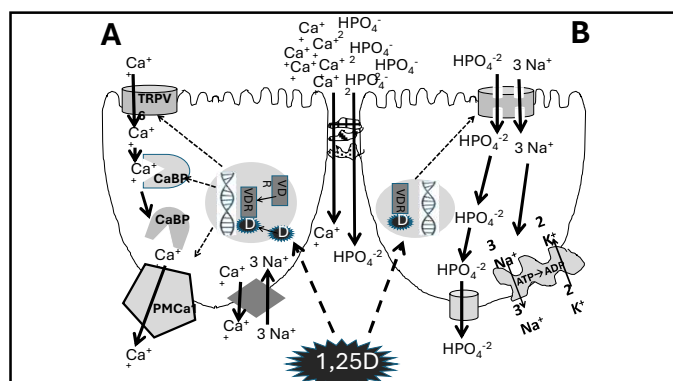
2. Diets High in **phosphorus** block enzyme producing 1,25 (OH)₂ vitamin D in kidney via **FGF23** made by bone cells.

Phosphorus Homeostasis

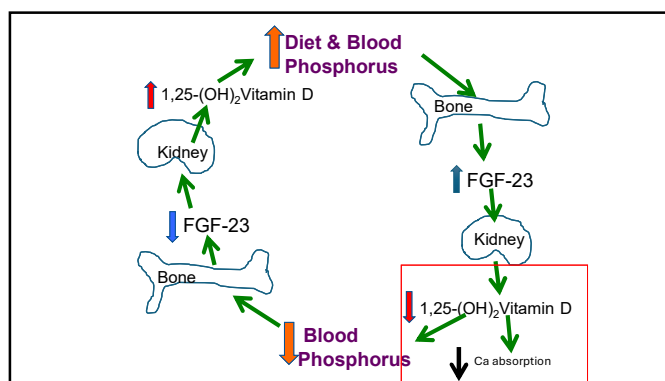


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1. Diets High in **potassium** causing alkalosis → bone & kidney PTH resistance

Anionic diets ↓ alkalinity of blood allowing parathyroid hormone to work!

2. Prepartum Diets High in **phosphorus** block enzyme producing 1,25 (OH)₂ vitamin D in kidney via FGF23.

Avoid Canola meal (1.05% P) and DDG (0.85% P)

Zeolite or Binder diets ↓ blood Phos, eliminating the inhibition of conversion of 25-OHD to 1,25(OH)₂ vitamin D

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1. Diets High in **potassium** causing alkalosis → bone & kidney PTH resistance

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2. Diets High in **phosphorus** block enzyme producing 1,25 (OH)₂ vitamin D in kidney via FGF23.

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3. Pre and Post Calving diets Low in **magnesium**- inhibits secretion and action of PTH

Add soluble Mg to diet! Get to 0.4% Mg diets

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Magnesium Metabolism

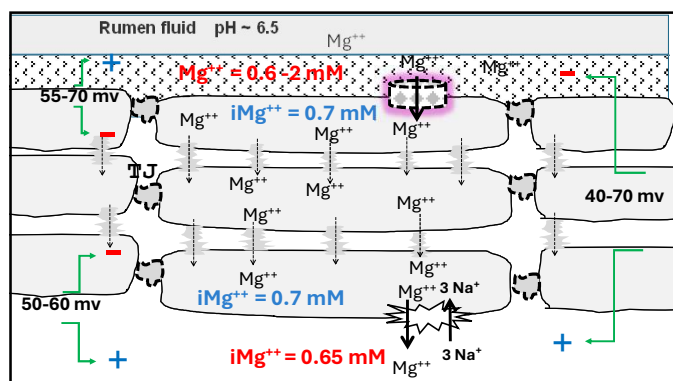
Diet Mg can only be absorbed across the rumen wall.

Only ionized (soluble) Mg can be absorbed.

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How Can Magnesium cross the Rumen wall???

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High K diets inhibit transport of Mg ions thru TRPM7 channel

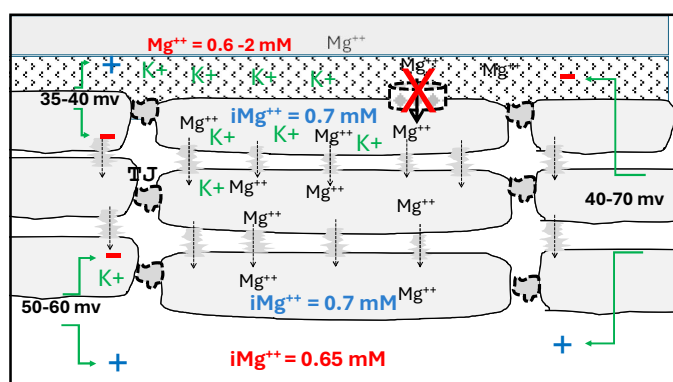
High potassium outside cell increases K⁺ leaking into cell (cell membranes are K leaky)

This causes potential difference across the apical membrane to be reduced

The driving force pushing Mg thru the TRPM7 channel is reduced!!!!

Mg⁺⁺ ion when diet Mg is low can no longer be transported by this mechanism

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Weiss (2004) calculated an average apparent digestibility for diet Mg, from all sources, of 24% when diet K was 1% (close to the K requirement for a dry cow) when diet Mg between 0.2 and 0.36%.

AND apparent digestibility of Mg was reduced by 7.5% for every 1% increase in diet K.

Tebbe, Wyatt, & Weiss (2018)- demonstrated bioavailability of MgO was considerably lower than NRC 2001 suggested

CORRECTED IN THE 2021 NASEM!!!!

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A 2nd Mg transporter exists at Apical Surface

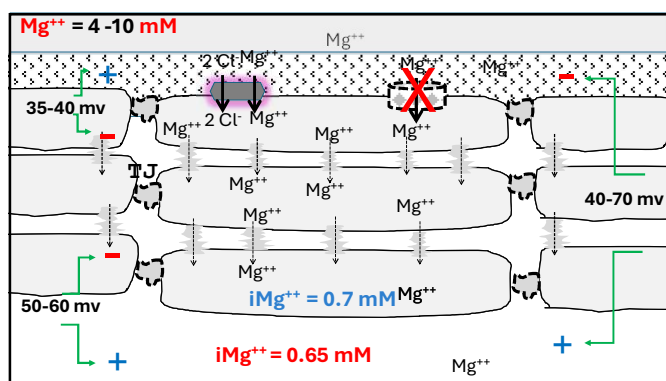
2Cl⁻ / Mg⁺⁺ Co-Transporter

Uses energy of Chloride diffusing into an area of lower concentration and Mg moving into an area of relative negativity to drive Mg from rumen fluid into the cells

But this mechanism requires much higher concentrations of Mg ions be present in the rumen fluid to work!!!

Pre-calving and post-calving diets with 0.4% Mg generally ensures this pathway will work

-mid-lactation milk fever!!

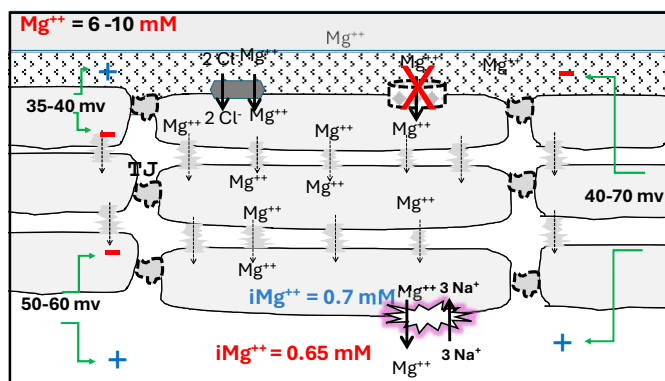


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Finally, the magnesium must be pumped across the basolateral side, AGAINST its electrical gradient, by a $\text{Mg}^{++}/3\text{Na}^{+}$ Exchanger

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Magnesium – **ONLY ABSORBED ACROSS RUMEN WALL**

Pre-calving

- using MgSO_4 or MgCl_2 as "anions" also supplies readily available, soluble Mg.

-The better anion supplements on the market include Mg in this form to remove Mg worries pre-calving.

-Zeolite diets – beware that the zeolite binds magnesium!!!

Post-calving is the bigger issue!!!!!!

Magnesium Oxide – supplies Mg and acts as rumen alkalizer.

MgO must become soluble to be available for absorption by rumen wall!!!!

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Testing Magnesium Oxide Availability

Weigh out 3 g MgO into large vessel.

Add 40 ml of 5% acetic acid (white vinegar) slowly!!

Cap container and shake well and let sit 30 minutes. Check the pH.

Vinegar will be pH 2.6-2.8!

The best MgO will bring the pH up to 8.2.

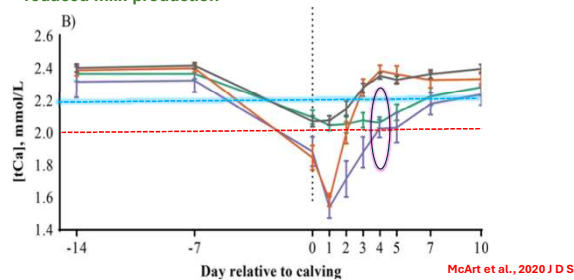
The worst to just 3.8.

pH is a log scale so this represents >10,000 fold difference in buffering action.

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Dyscalcemia = cows with blood Ca below 8.8 mg/dl (2.2 mM) on day 4.

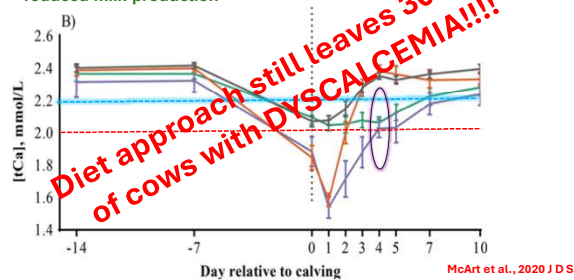
- reduced DM intake, rumination, fertility
- 1.7 to 5.3 times more likely to be culled or dead
- reduced milk production



77

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Oral Calcium Boluses

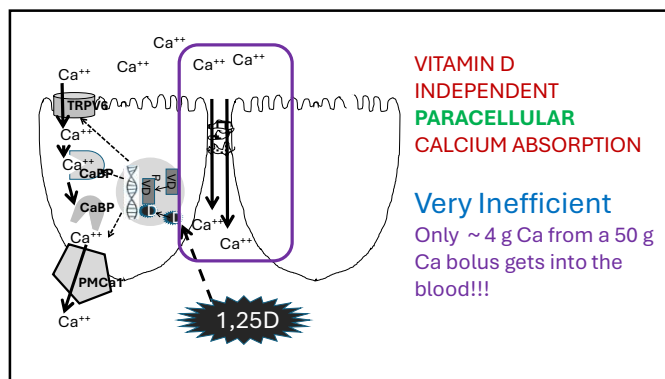
Typically each bolus supplies 30-45 g Ca in a highly soluble form, often as calcium chloride, to permit passive vitamin D independent absorption of Ca into the blood

- effect lasts JUST 3-6 hrs

Convenient, easy to use – give to multiparous cows at calving and often again 12-24 hours later

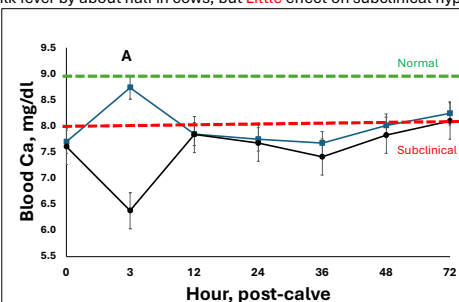
85% of dairies with less than 125 cows use Ca boluses
60% of dairies that feed an anionic diet use Ca boluses.

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Reduce milk fever by about half in cows, but **Little** effect on subclinical hypocalcemia



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Paracellular Ca Absorption

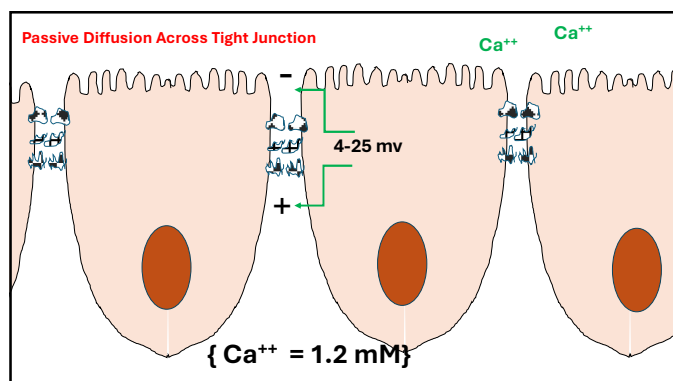
Ionized Ca concentration in interstitial fluids is ~ 1 mM.

To overcome the electrical potential stopping Ca^{++} from going across the tight junction requires a minimum of 4 mM Ca^{++} on luminal side of tight junction to get Ca^{++} to move to interstitial space.

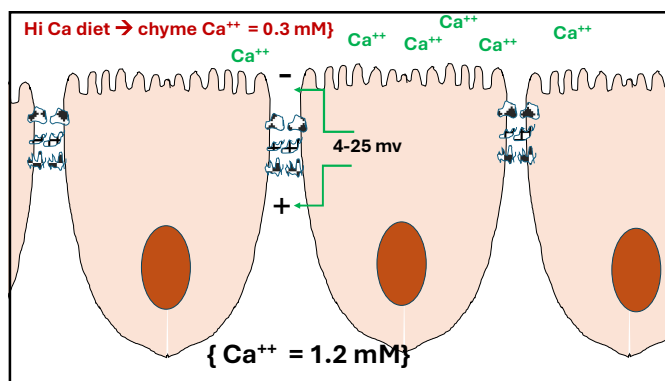
This can be achieved in calves fed milk (high Ca and very soluble) and in cows it is felt that with high Ca diets a portion (10-30%) of the Ca leaving the abomasum can be absorbed paracellularly in upper small intestine.

Ca Boluses- in localized areas of rumen the concentration of Ca^{++} will be high enough to push Ca^{++} across tight junctions for several hours

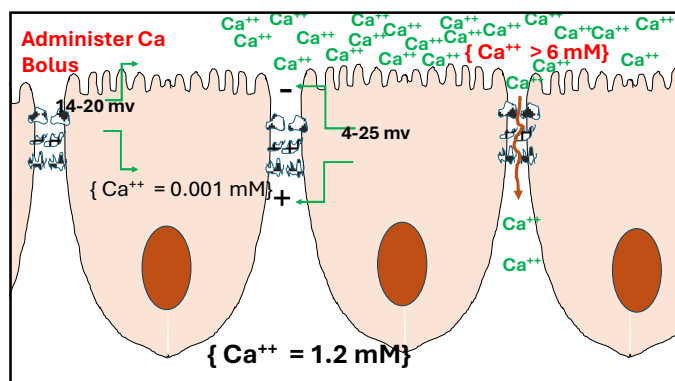
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Ca salts suitable for incorporation into boluses for rapid elevation of blood Ca must be highly soluble at rumen pH

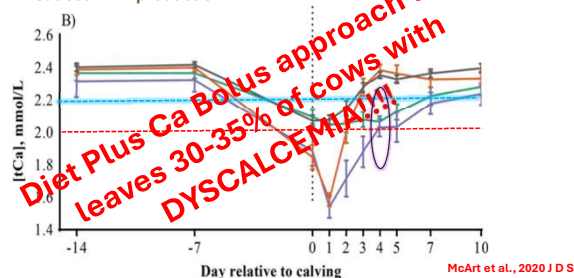
Solubility Within the Rumen (pH 6-7) at 39 Celsius

Calcium chloride = 104 g/ 100 ml water
 Calcium propionate = 38 g/100 ml water
 Calcium carbonate = 0.0007 g/ 100 ml water
 Calcium sulfate = 0.4 g/ 100 ml water
 Calcium lactate = 9.4 g/ 100 ml water
 Calcium acetate = 33 g/ 100 ml water
 Calcium gluconate = 5 g/100 ml water

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Dyscalcemia = cows with blood Ca below 8.8 mg/dl (2.2 mM) on day 4.

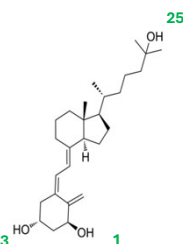
- reduced DM intake, rumination, fertility
- 1.7 to 5.3 times more likely to be culled or dead
- reduced milk production



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Solanum glaucophyllum

Plant that produces glycosides of 1,25-(OH)₂vitamin D₃.



Where OH on carbons 1,3 or 25 are replaced by 1-8 glucose molecules.

Glucose cleaved off in rumen liberating active form of vitamin D!!!

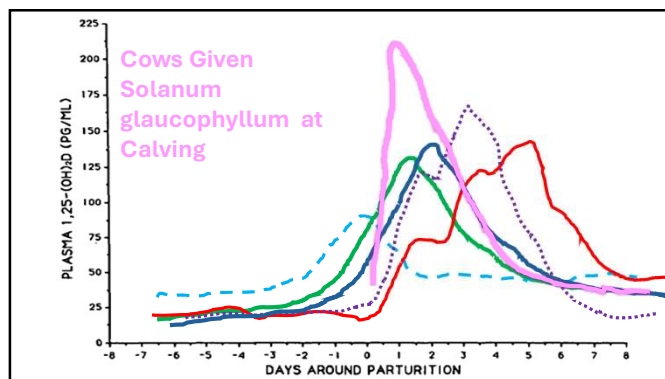


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Since failure to produce 1,25-dihydroxyvitamin D is a cause of Milk fever and hypocalcemia:

Can we utilize *S. glaucophyllum* to supply a natural bioactive form of vitamin D and help maintain more normal blood calcium concentrations in early days of lactation??

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New Zealand - 2 farms , one fed MgSO_4 , (urine pH 7.3) and one fed no anions at all
Cuttance et.al.,

Treatments - 43 cows/ treatment by Parity

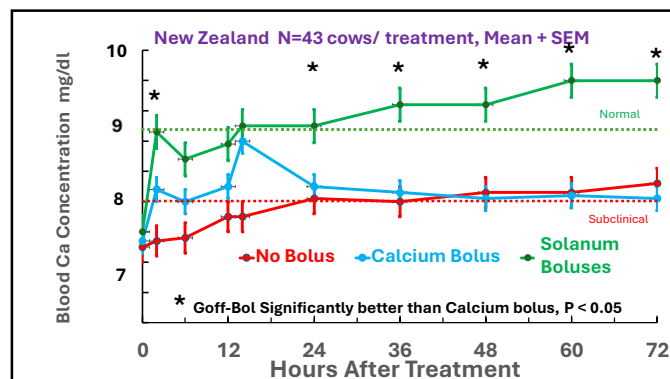
1. Two Solanum Boluses given at Time 0
2. Two Calcium boluses, one at time 0 and one at 12 hours
3. No Boluses

Blood sampled just before, and then 2, 6, 12, 14, 24, 36, 48, 60 and 72 hours after bolus administration (at either 7 AM or 7 PM)

8 mg/dl = cutpoint for subclinical hypocalcemia

8.8 mg/dl = cutpoint for Dyscalcemia

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Keys to reducing hypocalcemia and dyscalcemia in fresh cow

ACTIVATE TRANSCELLULAR CALCIUM TRANSPORT QUICKLY AFTER CALVING

Keep diet K and P low and diet soluble Mg elevated

Ensure the cow's GI tract is turned on by 1,25-dihydroxyvitamin D quickly

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