

P a g e : 1/10 Edition : 9 Revised Date : 12/5/21

# 1. IDENTIFICATION OF THE SUBSTANCE / PREPARATION AND OF THE COMPANY

**Product name** Mixture Ethylene Oxide 90% / Carbon Dioxide 10%

**Chemical formula**  $C_2H_4O + CO_2$ 

Uses Medical sterilization: chemical intermediate

Synonyms Ethylene Oxide: Oxirane; 1, 2-Epoxy ethane; EO; EtO; Ethene Oxide

Carbon dioxide: Carbonic acid gas; carbon anhydride

Company Balchem Sdn Bhd

No. 37, Lorong Sungai Puloh 1A/KU6,

Taman Teknologi Gemilang, Kaw. Perindustrian Sg. Puloh, 41050, Klang, Selangor, Malaysia T: +60 3 3290 2263 F: +60 3 3290 2016

E:info.asiapacific@balchem.com I: www.balchem.com

Emergency Contact CARECHEM 24

+61280144558

Malaysia 999

### 2. HAZARDS IDENTIFICATION

# 2.1 GHS classification

#### **Hazards Identification**

GHS02 – Extremely flammable gas	H220 / category 1
GHS04 – Gas under pressure: may explode if heated	H280 / liquefied gas
GHS06 – Toxic by inhalation	H331 / category 3
GHS08 – Irritating to eyes, respiratory system and skin	H335 / category 3
GHS08 – May cause cancer	H350 / category 1B
GHS08 – May cause inheritable genetic defects	H340 / category 1B



P a g e : 2/10 Edition : 9 Revised Date : 12/5/21

GHS classification

Hazard pictograms / symbols



# 2.2 Classified as dangerous according to the criteria of Regulation EC No 1272/2008

# Hazard Statement for physical hazards

H220 Extremely flammable gas.

H230 Chemically unstable gas (may react explosively even in the absence of air)

H280 Liquefied gas (may contain gas under pressure)

# **Hazard Statements for Health Hazards**

H302 Harmful if swallowed.
H315 Causes skin irritation.
H319 Causes serious eye irritation.
H331 Toxic if inhaled.

H335 May cause respiration irritation.
H340 May cause genetic defects.
H350 May cause cancer.

H372 Causes damage to the nervous system and to blood forming organs through prolonged or

repeated exposure by inhalation.

**Prevention** 

P202 Do not handle until all safety precautions have been read and understood.
P210 Keep away from heat/sparks/open flames/hot surfaces. No smoking.

P260 Do not breathe gas/vapours.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Response

P377 Leaking gas fire: Do not extinguish, unless leak can be stopped safely.

P304 + P340 IF INHALED: Remove to fresh air and keep at rest in a position comfortable for

breathing.

P308 + P313 IF exposed or concerned: Get medical advice/ attention.
P332 + P337 + P313 If skin or eye irritation occurs: Get medical advice/ attention.
P333 + P313 If skin irritation or rash occurs: Get medical advice/attention.



P a g e : 3/10 Edition : 9 Revised Date : 12/5/21

Storage

P405 Store locked up.

P410 + P403 Protect from sunlight. Store in well-ventilated place

### Other hazards

Carbon dioxide component is under high pressure but is generally non-toxic. Ethylene oxide is liquified gas at room temperature. Ethylene oxide is highly flammable even in this mixture with carbon dioxide. Beside hazard due to compressed gas of carbon dioxide, other hazards are due to ethylene oxide, which is highly flammable and generally toxic.

# 3. COMPOSITION / INFORMATION ON INGREDIENTS

## Substance/Preparation

Compressed gases mixture

#### Components/Impurities

Contains a mixture of liquified ethylene oxide and compressed carbon dioxide gas. Chemical toxicities are mainly due to ethylene oxide while carbon dioxide is generally non-toxic.

CAS Number	EC Number (from EINECS)	Name	%(Weight)
75-21-8	200-849-9	Ethylene oxide	90
124-38-9	200-696-9	Carbon Dioxide	10

### 4. FIRST AID MEASURES

## **General Information**

Take appropriate steps to avoid fire, explosion and inhalation hazards.

Remove contaminated soaked clothing immediately

Adhere to personal protective measures when giving first aid.

Seek medical treatment immediately.

#### Inhalation

Remove the casualty into fresh air and keep him immobile.

In the event of pulmonary irritation treat initially with corticoid spray, e.g. Ventolair or Pulmicort- metered-dose aerosol (Ventolair and Pulmicort are registered trademarks).

Seek medical treatment immediately.

In case of respiratory standstill give artificial respiration by respiratory bag (Ambu bag) or respirator.

Send for a doctor.



P a g e : 4/10 Edition : 9 Revised Date : 12/5/21

# Skin / eye contact

In case of frostbite spray with lukewarm (not hot) water for at least 15 minutes. Do not remove clothing frozen to the skin. Thaw it with lukewarm water. Apply a sterile dressing. Obtain medical assistance. Seek medical treatment immediately.

In case of contact with skin wash off immediately and for a long time (at least 15 minutes) with plenty of water. In case of eye contact, rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

Call for a doctor immediately.

### **Ingestion**

Ingestion is not considered a potential rote of exposure.

## 5. FIRE FIGHTING MEASURES

**Specific Hazards** Exposure to fire may cause containers to rupture/explode

Hazard combustion products Incomplete combustion may form carbon monoxide

Suitable extinguishing media All known extinguishers can be used

**Unsuitable extinguishing** Full water jet (MUST NOT be used for safety reason)

**Specific Methods** If possible. Stop flow of product

Continue water spray from protected position until container stays cool.

Spontaneous / explosive re-ignition may occur.

Extinguish any other fire.

**Protective equipment** Use self-contained breathing apparatus

Use chemically protective clothing.

# 6. ACCIDENTAL RELEASE MEASURES

**Personal precautions** Evacuate area

Use self –contained breathing apparatus and chemically protective clothing

Eliminate ignition sources

Wear self -contained breathing apparatus when entering area.

Ensure adequate air ventilation



# SAFETY DATA SHEET

# MIXTURE ETHYLENE OXIDE 90% CARBON DIOXIDE 10%

P a g e : 5/10 Edition : 9 Revised Date : 12/5/21

**Environmental precautions** Try to stop release

Prevent from entering sewers, and water systems as ethylene oxide is very

soluble in water.

Reduce vapor with fog or fine water spray.

Clean up methods Keep area evacuated

Ventilate area

Remove any source of ignition

Let any spill evaporate. Apply exhaust fan if possible.

Hose down area with water.

Wash contaminated equipment or sites of leaks with large quantities of water.

# 7. HANDLING AND STORAGE

**Handling** Use ONLY in thoroughly ventilated areas.

Ensure equipment is adequately earthed to prevent any spark from static

discharge.

Protect cylinder from any physical damage. Do not drop, drag, roll or slide

cylinder.

Do not let any water to be introduced into container with EtO.

Purge dry air into container and gas-flow system before introducing gas.

Use only specified equipment to handle this product with specified safe pressure

and temperature.

Refer to supplier's handling instructions.

Contact gas supplier if in doubt.

**Storage** Secure cylinders to the correct positioning.

Keep in dry well-ventilated area.

The gas cylinders must be adequately earth during storage. Segregate from oxidant gases and other oxidizing agents

Keep containers tightly closed and dry.

Storage condition: -

- In a cool, dry and well-ventilated area

- Away from direct sunlight. Protect from the heat (storage temperature  $\leq 50^{\circ}$ C)

- Away from any source of ignition (including static discharge)



Page: 6/10 Edition: 9 Revised Date: 12/5/21

# 8. EXPOSURE CONTROL / PERSONAL PROTECTION

**Exposure Limit value** Ethylene Oxide:

OSHA Time Weighted Average exposure limit (TWA): 1 ppm

OSHA Short Term Exposure Limit (STEL): 5 ppm

Carbon Dioxide:

OSHA Time Weighted Average exposure limit (TWA): 5,000 ppm

OSHA Short Term Exposure Limit (STEL): 30,000 ppm

**Engineering measure** Provide adequate general and local exhaust ventilation to maintain

concentrations below exposure and flammable limits.

**Personal protective equipment** Respiratory protection may not be required unless in case of leakage, use self –

contained breathing apparatus.

Use appropriate gloves and protective clothing for hand and skin protection.

Use safety glasses or ventilated goggle for eye protection.

A safety shower and eyewash station should be readily available.

For body and skin protection, wear coveralls, boots and/or other chemical

resistant protective clothing.

#### 9. PHYSICAL AND CHEMICAL PROPERTIES

Physical and chemical properties of the mixture as a product are not available. However, physical and chemical properties of each component are provided here as a guide.

Physical state at 20°C	High pressure liquefied gases mixture
Colour	Colourless gas mixture
Odour	CO <sub>2</sub> : Odourless
	EtO: sweet, ether-like smell
Flammability	EtO is extremely flammable and mixture with CO <sub>2</sub> is also flammable
Melting Point	CO <sub>2</sub> : -75.5 °C (sublimation point)
	EtO: -112 °C
Boiling Point	CO <sub>2</sub> sublime directly to gaseous form at -78.5 °C
_	EtO: 10.6 °C



Page: 7/10 Edition: 9 Revised Date: 12/5/21

Flash Point	CO <sub>2</sub> does not combust at any temperature. EtO: -18 °C (open cup)
Vapour pressure, 20°C	CO <sub>2</sub> is a gas at 20 °C EtO: 1.4 Bar
Relative Vapour Density	CO <sub>2</sub> : 1.5 (Air = 1) EtO: 1.52 (Air = 1)
Relative density, 20°C	EtO: $0.82$ (water = 1) CO <sub>2</sub> is a gas at 20 °C.
Solubility in water	CO <sub>2</sub> : 0.7 mole ratio at 20 °C and 1 atm pressure EtO: completely miscible in water (1,000 g/L)
Flammability range	CO <sub>2</sub> is not combustible EtO: 3 to 100 (vol % in air)
Auto-ignition temperature	CO <sub>2</sub> is not combustible EtO: 429 °C
Other data	Gases mixture is heavier than air. May accumulate in confined spaces, particularly at or below ground level.

### 10. STABILITY AND REACTIVITY

**Stability** EtO can form explosive mixture with air even in the presence of 10% CO<sub>2</sub>.

This EtO mixture may also decompose violently at high temperature and/or pressure in the presence at catalyst such as iron rust or other metal oxides

EtO may polymerize.

**Conditions to avoid** Air or oxygen

Water, humidity High temperatures

Materials to avoid Oxidizing agents, acids, organic bases, amines, ammonia and certain salts.

Reacts explosively with certain alcohols or mercaptans. Reacts with HCl to form highly toxic ethylene chlorohydrins. Avoid copper, silver, magnesium, mercury and their salts.



P a g e : 8/10 Edition : 9 Revised Date : 12/5/21

#### 11. TOXILOGICAL INFORMATION

Acute toxicity Ethylene Oxide

LD<sub>50</sub>/oral/rat:; 72 mg/kg

LC<sub>50</sub>/inhal/rat: 2.92 mg/1/15 min

1.44 mg / 1 / 4h

**Carbon dioxide**Relatively non-toxic

IDLH conc: 40,000 ppm (IDLH: Immediate Danger to Life and Health)

Skin Irritation/Sensitization Ethylene oxide is very irritating and may cause allergy reaction

Eye Irritation/Damage Ethylene Oxide cause serious eye irritation

Inhalation/Respiratory

**Sensitization Ethylene Oxide** cause respiratory tract irritation and may cause allergy reaction.

Carcinogenic risk to cancer by inhalation is well documented for EtO.

Carcinogenicity Ethylene Oxide IARC Classification Group 1: Carcinogenic to Human.

Carbon Dioxide is non-carcinogenic.

Mutagenicity Ethylene oxide has been shown to be an effective mutagen in a variety of

organisms from bacterial to mammals.

**Teratogenicity Ethylene oxide** is included in the Catalog of Teratogenic Agent by Thomas

Shepard based on one study. Medical counseling is recommended for exposure

of pregnant women to ethylene oxide.

Further information Studies showed that most toxicological impact exposure of EtO were through

inhalation. Exposure to a mixture of 90% EtO and 10% CO2 is still result is

substantial exposure to EtO.

### 12. ECOLOGICAL INFORMATION

General Ecological Impact Carbon Dioxide is a major green-house gas which contribute to Climate

Change and other ecological impact of Green-house gases. Other than that

carbon dioxide is not harmful to the environment.

Ethylene Oxide is toxic and harmful to the environment according to the

following criteria;



P a g e : 9/10 Edition : 9 Revised Date : 12/5/21

**Environmental Fate** Ethylene oxide is highly reactive. Thus, it does not persist indefinitely in the

environment. It is also very soluble in water. EtO will be converted gradually to

ethylene glycol in the environment.

**Aquatic Toxicity** LC<sub>50</sub> (Fat Minnow): 84 mg/L (96 hr exposure)

LC<sub>50</sub> (Daphnia magna): 137 – 300 mg/L (48hr exposure)

**Mobility** Atmosphere: Somewhat persistent in the atmosphere but will ultimately degrade

by process of photolysis.

Soil: Does not readily dissolved in soil and gradually will convert to ethylene

glycol.

Water: Very soluble in water but easily convert to ethylene glycol.

**Persistence/degradability** Readily undergoes biodegradation and hydrolysis in water and soil.

**Bio-accumulation** Does not bio-accumulate significantly.

# 13. DISPOSAL CONSIDERATION

Waste Disposal Do not dispose remaining gas to the environment. Return container and cylinder

to supplier.

**Additional Information Ethylene oxide** is also used as fumigant. Regulation of disposal of pesticides

and containers of pesticides may apply.

Do not attempt to refill cylinder with other gases or chemicals.

### 14. TRANSPORT INFORMATION

Proper shipping name Ethylene Oxide and Carbon Dioxide mixture

UN Number UN3300

Hazard Class 2 – Flammable Gas

ADR/RID Classification Code 2TF ADR/RID Hazard Nr 263 Packing Group None

Label 2.3: Toxic substance Label 2.1: Flammable gas

IMDG EmS codesF-D, S-UIMDG Marine pollutantNoPassenger AircraftForbidden



P a g e : 10/10 Edition : 9 Revised Date : 12/5/21

SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure position, in a well-ventilated vehicle. Cylinders transported in an enclosed, non-ventilated compartment of vehicle can present serious safety hazards.

### 15. REGULATORY INFORMATION

Department of Occupational Safety and Health, Ministry of Human Resources, Malaysia Industrial Code of Practice on Chemicals Classification and Hazard Communication, 2014.

Chemical Name: ETHYLENE OXIDE

CAS No. 75-21-8

Classification H-Code Classification Code
H280/281 Pressurized Gas

H280/281 Pressurized Gas H220 Flam. Gas 1 H350 Car. 1B H340 Muta. 1B

H331 Acute Tox. 3 (inh) H319 Eye Irrit. 2 H335 STOT SE 3

H335 STOT SE 3 H315 Skin Irrit. 2

Chemical Name: CARBON DIOXIDE

CAS No. 124-38-9 Classification Not listed

# **16. OTHER INFORMATION**

The information and opinion presented herein are based on data and scientific information currently available. Since the use of information in this Safety Data Sheet and the conditions of use are not within the control of Balchem Sdn Bhd., it is the user's obligation to determine the conditions for safe use of this product.

Ensure all national and local regulations are observed. Ensure operators and handlers of this product understand the hazard and toxicological effect from accidental and unnecessary exposure.

While proper care has been taken in the preparation of this document, no liability for injury or damage resulting from its use can be accepted by the manufacturer.

Edited by: Dr. Md. Sani Ibrahim (PhD in Organic Chemistry)

e-mail: sani@sanichem.com.my

Date: 15 January 2019